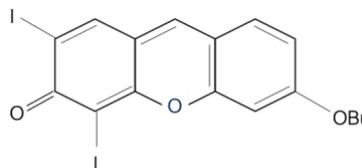


Product Description

Name: 5,7-diiodo-3-butoxy-6-fluorone
 CAS # 161728-47-8
 Formula: C₁₇H₁₄I₂O₃
 Synonyms: DIBF, H-Nu 470



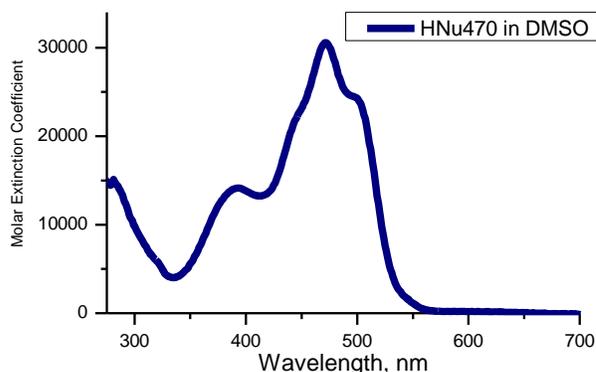
- Initiates both free-radical and cationic photopolymerization, using a different set of co-initiators depending on the polymerization process
- Type II photoinitiator: H-Nu 470 acts as a light harvester, followed by energy or electron transfer to a co-initiator molecule; produced reactive species activate photopolymerization
- Commercial photoinitiator (TSCA status: LVE), non-toxic (LD50>5000 mg/kg)

Advantages

- ✓ High absorptivity, low concentrations are needed (0.01-0.15% by weight)
- ✓ Capable of significant depth of cure > 1 inch
- ✓ Time and energy savings when one-pass thick cure can replace thin multilayered coatings
- ✓ Initiator bleaching (bright red to pale orange/no color) upon light interaction can act as cure indicator

Typical Properties

Property	Test Method	Typical Value
Absorbance Maximum	UV Visible spectroscopy	470 nm
Appearance	Visual	Orange Powder
Melting point	Melting point Analysis	>270 °C
Molar Extinction Coefficient	UV Visible spectroscopy	30,200 (at 470nm)



Applications Recommendations

- ✓ Direct solubility of H-Nu 470 in resins can be difficult, pre-dissolution in one of the following resins/solvents before adding resin is recommended:

Free-Radical Diluents						Cationic Diluents				
DMAA	VMOX	THFA	HDDA	HPMA	IBOA	Heloxy61	Heloxy62	EOXA ¹	ECC ²	Eponex1510
S	S, T°	S, T°	I	P, T°	I	I	S	S, T°	S, T°	S, T°

1% of H-Nu 470 in diluent is: S-soluble T°-requires heat (60°C for several hours) P-partially soluble I-insoluble

1 = 3-ethyl-3-hydroxymethyloxetane

2 = 3,4-epoxycyclohexylmethyl-3,4

Note: DMAA is an excellent solvent for H-Nu 470, but it **cannot** be used in **cationic polymerization** as it inhibits process.

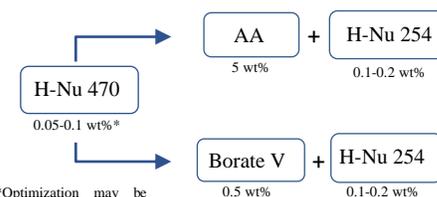
- ✓ The presence of any undissolved red particles is an indication of incomplete solubility. More heating (60°C) or stirring may help with incorporation, or pre-dissolution using different diluents is recommended.

Polymerization mechanisms

Free-Radical Cure (Acrylates)

- ✓ H-Nu 470 concentration range spans from 0.05 to 0.2 wt.%, depending on a cure thickness. 0.05 wt% is recommended starting point for thick cure (>1mm) and 0.1wt% for thin cure.
- ✓ Amine acrylates (AA) are needed as co-initiators at 5-10 wt.%. If amine use is counterindicated (acidic adhesion promoter, outgassing), as an alternative, another co-initiator, Borate V (0.5 wt.%) may provide similar cure response as typical amine co-initiators
- ✓ H-Nu 254 iodonium salt is recommended for acceleration.

Recommended starting concentrations:

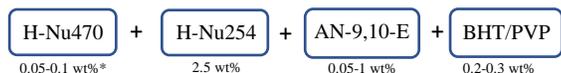


*Optimization may be necessary for each individual application

Cationic Cure (Epoxides)

- ✓ H-Nu 470 concentration range spans from 0.05 to 0.2 wt.%, with a good starting point at 0.1 wt.% based on solids.
- ✓ H-Nu 254 iodonium salt is needed to achieve cure in epoxide resins (sulfonium salts will not work).
- ✓ Accelerator AN-9,10-E can greatly enhance cure speed and sensitivity and is needed to achieve cure.
- ✓ Amines or DMAA solvent cannot be used due to the superacid “poisoning” effect.
- ✓ Caution & experimentation is required at higher temperatures as it can lead to thermal ground state polymerization. Additional stabilizers, depending on the photopolymerization matrix can be used. As an example, 0.2 wt.% of PVP+BHT* mixture (1:1) can be added to glycidyl epoxide and 0.3 wt.% of PVP+BHT mixture (1:1) to cycloaliphatic epoxide.

Recommended starting concentrations:



*Optimization may be necessary for each individual application
PVP = Polyvinylpyrrolidone
BHT = Butylated hydroxytoluene

Examples of H-Nu 470 uses:

Coating process

- Moore, M., Lungu, V., Marino, T., Radtech Report 11, 2, (March/April) 1997

SU-8 photoresist

- Y. Lin, P.R. Hermann, and K.Darmawikarta, Appl. Phys. Lett. 86, 7, 071117 (2005)
- J.H. Moon, S.-M. Yang, D.J. Pine, and W.-S. Chang, Appl. Phys. Lett. 85, 18, 4184 (2004)
- D. Rodriguez Ponce, K Lozano, et al. J. Polym. Sci.: Part B: Polym. Phys. 48, 1, 47 (2010)

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- D. Ahn, L.M. Stevens, K. Zhou, and Z. A. Page, ACS Cent. Sci. 6, 9, 1555-1563 (2020)

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